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54 Drug particles having constant release.

57 Drug particles are disclosed which reduce the variation in dissolution rate with time frequently found with sustained-release dosage forms. This is achieved by coating the drug particle, which may contain a surfactant and a disintegrant in addition to the drug active, with a first coating of a water soluble polymer, a second coating of an enteric material and a third coating of a material which is insoluble at any pH and acts as a diffusion barrier for the drug active.

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DRUG PARTICLES HAVING CONSTANT RELEASE

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TECHNICAL FIELD

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The present invention is related to sustained release drug particles having a release profile providing for more constant release of the drug active.

BACKGROUND OF THE INVENTION

10 This invention relates to pharmaceutical particles providing a prolonged release of the drug active.

The administration of drugs orally oftentimes presents serious problems. In very many cases, it is necessary for the drugs to be absorbed from the gastro-intestinal tract into the
15 bloodstream. When a drug is given by mouth, the assimilation is normally fairly rapid and the level in the blood reaches a maximum. Then it declines as the drug is excreted or otherwise removed from the bloodstream. The blood level falls and finally reaches a level so low that it is no longer effective and it is then
20 necessary to take another dose. The peaks and valleys thus caused in blood level have several disadvantages. One is the necessity of taking frequent small doses in order to maintain a therapeutically desired blood level. This is awkward and presents a problem as the patient has to continue to take further
25 doses. A second drawback is that in order to have a therapeutically useful blood level, each dose has to be fairly large so that after the blood level peak is reached and begins to subside, there will still remain sufficient of the drug to be useful. The necessity for large doses brings with it some
30 problems. The large initial doses may be unacceptable or may produce some undesirable side reactions.

For the reasons set out above, various attempts have been made to slow up the release of drugs taken orally, producing a sustained release product. Sometimes it is felt desirable only to
35 slow up the release sufficiently so that it extends over a long enough time to even out the more marked peaks and valleys in blood level. In other cases, it may be desirable to prevent

assimilation from the stomach but to permit assimilation from the intestine. Various coatings have been developed (i.e., enteric coatings) to help achieve this purpose.

5 Some of the attempts at making improved sustained-release particles are reflected in patents issued in the area. One patent is U.S. 3,078,216, February 19, 1963 to Greif, which discloses sustained-release particles having a coating of water-insoluble wax and possibly, a second enteric coating. U.S. Patent 4,459,279, July 10, 1984 to Stricker et al., discloses drug granules having a
10 coating of a water insoluble component and a water soluble component. As a final second coating an enteric polymer may be employed. European Patent Application 0 061 217, September 29, 1982 to Giudice et al., discloses drug particles having an inert core, a first layer of the drug and an outer second layer of
15 polyvinyl pyrrolidone. European Patent Application 0 092 060, October 26, 1983 to Colombo et al., discloses drug particles having an enteric polymer in the drug core, a first coating designed to regulate the release of the drug and a second coating to protect the first coating.

20 While the above described references disclose delayed release drug particles, they as well as other prior art executions, do not provide complete answers to making totally satisfactory delayed release products, particularly those containing propionic acid derivatives.

25 Some drugs, such as ibuprofen, do not appear to follow a pattern of constant release found for other drugs. Instead, the dissolution data for ibuprofen particles having a porous coating indicate that the controlling surface area is not the constant surface area of the porous coating, but the surface area of the
30 ibuprofen particle which decreases with time as the drug dissolves. Thus, the dissolution rate decreases with time.

The decrease in dissolution rate with time of an ibuprofen particle having a porous coating can be compensated for if the resistance of the porous coating to diffusion also decreases with
35 time. This can be achieved by decreasing its thickness with

time. The present invention provides a means of producing a coating on an ibuprofen particle that decreases in thickness at a rate such that the decrease in surface area of the ibuprofen particle as it dissolves is compensated for and the dissolution rate remains essentially constant.

It is a further object of the present invention to provide an analgesic effective against a variety of pain, acute and chronic.

It is still a further object of the present invention to provide delayed release particles which are effective in the treatment of dysmenorrhea.

These and other objects will become readily apparent from the detailed description which follows.

As used herein all percentages and ratios are by weight unless otherwise specified.

SUMMARY OF INVENTION

The present invention relates to therapeutic particles containing a drug active core, a first coating of a water soluble polymer, a second coating of an enteric material and a third coating of a material which is insoluble at any pH and acts as a diffusion barrier for the drug active.

DETAILED DESCRIPTION OF THE INVENTION

The essential as well as optional components of the present particles are described below. In the present application the following terms have the meanings given.

"Pharmaceutically-acceptable" or "pharmacologically-acceptable", as used herein, means that the ingredients used in the compositions are suitable for use in contact with the tissue of humans, without undue toxicity, irritation, allergic response, and the like, commensurate with a reasonable benefit/risk ratio.

The term "comprising", as used herein, means that various other compatible components, including both active and inert ingredients, can be conjointly employed in the compositions of this invention. The term "comprising" thus encompasses and includes the more restrictive terms "consisting of" and "consisting essentially of".

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By "compatible" herein is meant that the components of the present invention are capable of being commingled without interacting in a manner which would substantially decrease the efficacy of the drug active under ordinary use conditions.

5 Drug Active

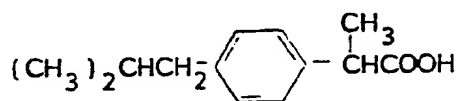
The drug actives found useful in the particles of the present invention are propionic acid derivatives. Such compounds have both analgesic and anti-inflammatory activity and can be used at appropriate dosage levels in the present invention. The compounds are frequently used in their free acid form but also

10 can be used in the form of their pharmaceutically accepted salts. The propionic acid derivatives for use herein include, but are not limited to, ibuprofen, flurbiprofen, fenoprofen, ketoprofen, indoprofen, suprofen, and fluprofen. Structurally related propionic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group. Structural formulas for representative group

15 Propionic Acid Derivatives

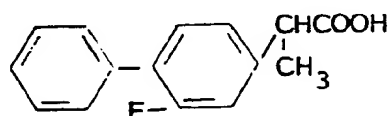
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ibuprofen



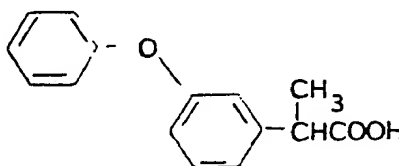
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flurbiprofen



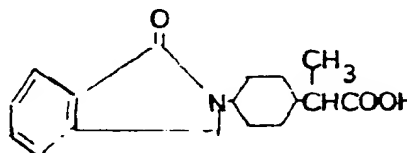
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fenoprofen

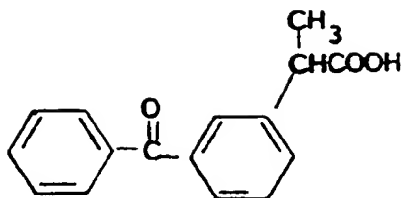


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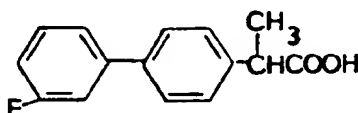
indoprofen



ketoprofen



fluprofen



Thus, "propionic acid derivatives" as defined herein are non-narcotic analgesics/nonsteroidal anti-inflammatory drugs having a free $--CH(CH_3)COOH$ or $--CH_2CH_2COOH$ group (which optionally can be in the form of a pharmaceutically acceptable salt group, e.g., $--CH(CH_3)COO-Na^+$ or $--CH_2CH_2COO-Na^+$), typically attached directly or via a carbonyl function to a ring system, preferably to an aromatic ring system.

A drug preferred for use in the present compositions is ibuprofen.

Preferably included with the propionic acid derivative in the core of the present particles is a surfactant. A preferred surfactant is polysorbate 80 (POE [20] sorbitan monooleate) which helps in allowing the drug to be wetted out and more easily pass through the coatings. The surfactant is used at a level of from about 0.05% to about 2%, preferably from about 0.1% to about 1% based on the weight of the total core solids.

Another material preferably included with the propionic acid is a disintegrant. A preferred material is a polyvinylpyrrolidone, PVP XL. This material, a crosslinked homopolymer of N-vinyl-2 pyrrolidone, being hydrophillic, serves as a disintegrant for the propionic acid derivative. The polymer is used at a level of from about 0.2% to about 4.0%, preferably from about 0.5% to about 2.5%, based on the weight of the total core solids.

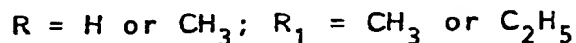
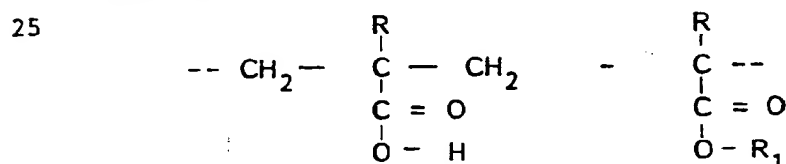
The size of the core composition can be chosen to meet the formulator's needs but is preferably from about 0.1mm to about 2mm, most preferably from about 0.5mm to about 1.2 mm.

First Coating

The therapeutic cores of the present particles have a first coating of a water soluble polymer. A preferred material is a polyvinylpyrrolidone polymer, PVP K-30, having a molecular weight of about 40,000. Other materials such as PVP K-90 having a molecular weight of about 360,000, PVP K-15 having a molecular weight of about 10,000 or hydroxypropyl methylcellulose (HPMC) may also be used. The polymer is present at a level of from about 0.2% to about 5%, based on the weight of solids in the core and the first coating. Preferred levels are from about 0.5% to about 2.0% for PVP K-30, about 0.5% to about 1.8% for PVP K-90, about 1.0% to about 4.0% for PVP K-15 and about 0.4% to about 1.3% for HPMC. If desired some of the propionic acid derivative or a surfactant may be incorporated into this coating.

15 Second Coating

The second coating used on the present particles is an enteric polymer. The polymer is selected from the group consisting of acrylic polymers and copolymers. The material should not be soluble in the stomach (i.e., acidic pH environment) but be soluble in the intestines. A preferred material is EudragitTM L30D offered by Rohm Pharma. This material is a copolymer, anionic in character, based on polymethacrylic acid and acrylic acid esters having the following repeating structure:



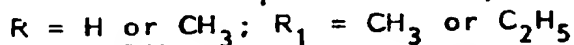
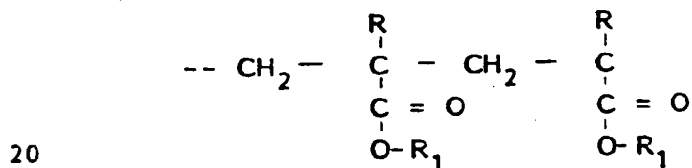
30 The L30D material has a mean molecular weight of about 250,000.

The second coating is preferably applied to the core so that the weight of coating solids is from about 1% to about 30% based on the total weight of the core plus first coating solids and second coating solids, preferably from about 2% to about 20%.

Materials which modify the enteric coating can be used with the enteric. Such materials include pharmaceutically acceptable plasticizers such as triacetin which are compatible with the drug active. Plasticizers are used at a level of from about 0% to about 25%, preferably from about 0% to about 15% based on the weight of the solids in the first coating.

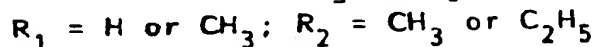
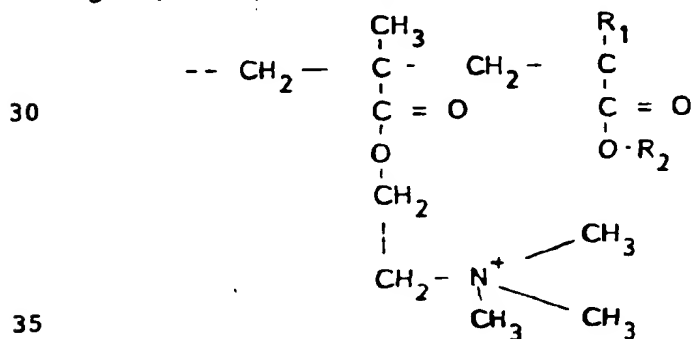
Third Coating

The third coating on the present particles serves as a diffusion barrier for the drug active. The coating material is a polymer or copolymer based on methacrylic acid. The material should be permeable to water and solubilized drugs but is preferably not soluble at any of the pHs found in the stomach or the intestines. A preferred material is EudragitTM E 30D offered by Rohm Pharma. This material is a copolymer, neutral in character, based on poly(meth)acrylic acid esters having the following repeating structure:



EudragitTM E-30D has a mean molecular weight of about 800,000

Materials which can be used as a partial replacement of Eudragit E-30D are copolymers synthesized from acrylic and methacrylic acid esters with a low content of quaternary ammonium groups. The repeating structure is as follows:



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The molar ratio of the ammonium groups to the remaining meth(acrylic) acid esters is 1:20 with a material identified as EudragitTM RL and 1:40 with a material identified as EudragitTM RS.

5 As with EudragitTM E-30D material, the RL and RS materials are permeable to water and to the drug actives but are virtually insoluble at any pH.

As with the second coating, the third coating may contain pharmaceutically acceptable materials which modify the coating.
10 Such materials include hydroxypropyl methyl cellulose, talc, sodium chloride, and polysorbate 80. These materials are conventional adjuvant materials for use with coatings and can be used at levels found by the formulator necessary to give the effect desired (e.g., hydroxypropyl methyl cellulose up to about
15 8% of the acrylic polymer; talc up to about 35% of the acrylic polymer; polysorbate 80 up to about 4% of the acrylic polymer; and sodium chloride up to about 10% of the acrylic polymer).

The third coating is used in the present particles at a level such that the coating is from about 2% to about 20% of the total
20 solids in the core, the first coating, the second coating and the third coating, preferably from about 4% to about 8%. The amount of acrylic polymer in the third coating, on this same basis should be from about 1.2% to about 11.8%, preferably from about 2.4% to about 4.7%.

25 METHOD OF MANUFACTURE

A method for manufacturing the particles of the present invention is given in Example I.

FIELD OF USE

30 The particles of the present invention can be used in tablets, capsules or any other convenient form. As indicated previously, the drug actives are known for providing relief from a wide variety of pain.

The following examples further describe preferred embodiments within the scope of the present invention. They are given
35 solely for illustration and are not to be construed as limitations of this invention as many variations thereof are possible without departing from the invention's scope and spirit.

EXAMPLES I-IV

The following are representative particles of the present invention.

		<u>Weight %</u>			
5	<u>Component</u>	<u>I</u>	<u>II</u>	<u>III</u>	<u>IV</u>
	Ibuprofen	87.10	89.5	58.45	58.45
Core	Polysorbate 80	0.44	--	0.3	0.3
	PVP XL	1.79	--	0.3	0.3
10	First Coating PVP K-30	0.90	1.1	--	0.6
	Hydroxypropyl methylcellulose	--	--	0.6	--
	Ibuprofen	--	--	29.6	29.6
	Polysorbate 80	--	--	0.07	0.07
15	Second Coating Eudragit TM L 30D	5.77	5.4	5.2	5.2
	Triacetin	--	--	0.5	0.5
20	Third Coating Eudragit TM E 30D	4.0	4.0	2.9	2.9
	Talc	--	--	1.7	1.7
	Polysorbate 80	--	--	0.08	0.08
	Hydroxypropyl methylcellulose	--	--	0.3	0.3
		<u>100.0</u>	<u>100.0</u>	<u>100.00</u>	<u>100.00</u>

25

The particles described above are prepared by first compacting the ibuprofen, polysorbate 80 and PVP XL at 115 psi to 195 psi, preferably from 145 to about 165 psi, in a roller compactor. The compaction is then milled and the resulting particles sieved to obtain particles in the range of 0.1mm to about 2.0mm, preferably from about 0.5mm to about 1.2mm.

The PVP K-30 is applied to the uncoated cores by means of a centrifugal granulator operated at a temperature of from about 20-40°C. The particles with this first coating are removed and dried overnight in a 50°C oven.

35

The cores are then coated with the EudragitTM L 30D coating followed by the EudragitTM E 30D coating. An L 30D coating dispersion is prepared by diluting the coating solids with water to obtain a dispersion having about 16.5% non-volatiles.

5 The cores coated with the water soluble polymer are then introduced into a Wurster fluidized bed coater using fluidizing air having a temperature of from about 35 to 60°C. The coating dispersion is next added to the coater by means of a peristaltic pump. The atomizing and fluidizing air flow rates and coating
10 dispersion are adjusted to maintain a coating temperature of 25 to 40°C.

The cores with the L 30D coating are subsequently coated with a dispersion of the EudragitTM E 30D. The dispersion having a solids concentration of about 20% is introduced into the
15 coater using the same conditions used to apply the L30D coating. The particles at this point are then cured for a number of hours at elevated temperatures.

The coated particles exhibit excellent release of the ibuprofen over a period of many hours.

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CLAIMS

1. Drug particles comprising:
 - (a) a propionic acid derivative core;
 - (b) a first coating of a water soluble polymer;
 - (c) a second coating of an enteric material selected from the group consisting of acrylic polymers and copolymers; and
 - (d) a third coating of a material insoluble in the stomach and intestines selected from the group consisting of methacrylic acid polymers and copolymers.
2. Drug particles according to Claim 1 wherein the propionic acid derivative is selected from the group consisting of ibuprofen, flurbiprofen, fenoprofen, ketoprofen, suprofen, indoprofen and fluprofen.
3. Drug particles according to Claim 2 wherein the first coating is present at a level of from about 0.2% to about 5% of the total weight of the core and the first coating.
4. Drug particles according to Claim 3 wherein the second coating is present at a level of from about 1% to about 30% of the total weight of the core, the first coating and the second coating.
5. Drug particles according to Claim 4 wherein the third coating is present at a level of from about 2% to about 20% of the total weight of the core, the first coating, the second coating and the third coating.
6. Drug particles according to Claim 5 wherein the propionic acid derivative is ibuprofen and a surfactant and a disintegrant are included with ibuprofen in the core.
7. Drug particles according to Claim 6 wherein the first coating is a polyvinylpyrrolidone polymer.
8. Drug particles according to Claim 7 wherein the second coating is based on polymethacrylic acid and acrylic acid esters and has a mean molecular weight of about 250,000.
9. Drug particles according to Claim 8 wherein the third coating is based on polymethacrylic acid esters and has a mean molecular weight of about 800,000.